THE PREPARATION AND CRYSTAL STRUCTURE OF THE SALT $[C_6F_5NWF_4$ -F-WF₄-NC₆F₅]⁻C₆F₅ $\dot{N}H_3$

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The reactions of amines with tungsten hexafluoride to yield [W(NR)] species have been well documented and studied by means of ¹⁹F n.m.r., but, to date, have not yielded crystal structures. Kokunov <u>et al</u>. (Koord. Khim., 1975, 1(8), pp. 1100-5) first postulated the formation of imido-tungsten fluoride species from ¹⁹F n.m.r. data, obtained from acetonitrile (MeCN) solutions of hexafluoride and amine mixtures. The data established the formation of both neutral and anionic species; $WF_4(NR)$.MeCN and $WF_5(NR)$ NRH_3 (R = H, Bu^t). More detailed work by Chambers <u>et al</u>. (J. Chem. Res. (M), 1977, pp. 1846-73) supported this idea, and also indicated the formation of an anionic, dimeric species, $[MeN=WF_4-F-WF_4=NMe]^-$, with linear C-N-W-F-W-N-C framework. However, to date, no crystal structures of this type of compound have been obtained. By treating WF_6 with the bulky amine pentafluoroaniline (C₆F₅NH₂) a yellow, crystalline solid is formed, which has a mass balance corresponding to the formation of a 3:1 adduct

$$WF_{6}(xs) + C_{6}F_{5}NH_{2} \rightarrow WF_{6}.3C_{6}F_{5}NH_{2}'$$

The solid dissolves readily in MeCN, anhydrous HF and is less soluble in trifluoroacetic acid (TFA). Crystals grown from TFA proved suitable for X-ray studies, and were found to be triclinic, space group Pl. The unit cell is occupied by two formula units of the dimeric species $[C_6F_5N=WF_4-F-WF_4=NC_6F_5]^-C_6F_5NH_3$, each with a near linear C-N-W chain but with a slightly differing geometry about the bridging fluorine. Terminal fluorine atoms bonded to tungsten are bent away from the bulky $-NC_6F_5$ moieties. Principal bond distances and angles are: $W-F_{bridge}(average)$, 2.15(4); $W-F_{terminal}(average)$, 1.92(3); W-N(average), 1.78(4); N-C(average), $1.45(4)A^2$; $W_1-F_1-W_2$, $150.73(1.3)^\circ$; $W_3-F_{31}-W_4$, $169.43(2.0)^\circ$.